

Tesi doctoral presentada per En/Na

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amb el títol

"Preparation, characterization and modeling of zeolite NaA membranes for the pervaporation dehydration of alcohol mixtures"

per a l'obtenció del títol de Doctor/a en

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A:	Work of adsorption according to Eq. I.12 [J mol^{-1}]; Parameter in Eq. III.2 [$\text{mol m}^{-2} \text{s}^{-1} \text{Pa}^{-1}$]; Area drawn by breakthrough curves in Eqs. III.4 and A.1-A.2 [s^{-1}]; Parameter in the Antoine Equation; Pre-exponential factor in Eq. I.40 and VIII.45
A_{12}, A_{21} :	Parameters of the Van Laar model (Eq. VII.35)
A_b :	Bed cross section [m^2]
a:	Activity [-]
a_m :	Packing factor [$\text{m}^2 \text{m}^{-3}$]
B:	Parameter in Eq. III.2 [$\text{mol m}^{-2} \text{s}^{-1} \text{Pa}^{-2}$]; Parameter in the Antoine Equation
B_{ij}^S :	Factor defined according to Eq. I.45 [s m^{-2}]
C:	Concentration
d or d_p :	Pore size [m]
\bar{d}_i :	Mean pores of the function $\varepsilon(d)$ [m] defined by Eq. VI.3
\bar{d}_p :	Mean pore size [m]
$d_{m,i}$:	Kinetic diameter of the species i [m]
$\langle d^n \rangle$:	n^{th} moment of the function $\varepsilon(d)$ [m^n]
D:	Fick or transport diffusivity [$\text{m}^2 \text{s}^{-1}$]
D^* :	Self-diffusivity [$\text{m}^2 \text{s}^{-1}$]
D_0 :	Pre-exponential factor according to Eq. I.34
D_b :	Bed diameter [m]
D_{ij} :	Molecular diffusivity between species i and j
D_L :	Viscous diffusivity [$\text{m}^2 \text{s}^{-1}$]
D_p :	Particle size [m]
D_∞ :	Solute diffusion in an environment without interaction with the pore walls
D_\pm :	Electrolyte diffusion in an environment without interaction with the pore walls
\mathfrak{D}_i^S :	Maxwell-Stefan surface or “jump” diffusivity of species i [$\text{m}^2 \text{s}^{-1}$]
\mathfrak{D}_{ij}^S :	Maxwell-Stefan counterexchange diffusivity of species i and j [$\text{m}^2 \text{s}^{-1}$]
E:	Surface energy in Eqs. I.4 and VII.27 [J mol^{-1}]
E_D :	Activation energy for diffusion [J mol^{-1}]
E_{eff}^S :	Effective activation energy [J mol^{-1}]
E° :	Characteristic energy in the Dubinin-Astakhov isotherm (Eqs. I.14 and VII.16)
$E(d^*)$:	Cumulative pore size function
f:	Fugacity [kPa]
F_o :	Fourier number [-]
\bar{G} :	Gibbs free energy [J kg^{-1}]
G° :	Dimensionless integral free energy at molar saturation loading in Eq. 15 [-]
He:	Henry coefficient [$\text{mol kg}^{-1} \text{kPa}^{-1}$],
IEP:	Isoelectrical point [-]
j_b :	Colburn factor [-]

k:	Parameter of the Potential Thermodynamic Isotherm (Eq. VII.15)
k_B :	Boltzmann's constant [$1.38 \times 10^{-23} \text{ J K}^{-1}$]
k_G :	External mass transfer coefficient [m s^{-1}]
ℓ :	Membrane thickness [m, μm]
L_b :	Bed length [m]
M:	Molecular weight [kg mol^{-1}]
\bar{M} :	Mean molecular weight [kg mol^{-1}]
m:	Weight [kg]; Exponent in the Potential Thermodynamic Isotherm (Eq. VII.15)
MWCO:	Molecular weight cutoff [kg mol^{-1}]
n:	Exponent of the Freundlich isotherm (Eq. I.8) [-]; Number of mols [mol]
N:	Flux [$\text{kg m}^{-2} \text{ h}^{-1}$; $\text{mol m}^{-2} \text{ s}^{-1}$]; Permeance [$\text{mol m}^{-2} \text{ s}^{-1} \text{ Pa}^{-1}$]; Number of finite elements [-]
N_A :	Avogadro Number [$6.023 \times 10^{23} \text{ mol}^{-1}$]
N_{Pe} :	Peclet Number [-]
N_{Sc} :	Schmidt Number [-]
N_{Sh} :	Sherwood Number [-]
P:	Pressure [Pa]
P_m :	Mean pressure [Pa]
P^0 :	Saturation vapor pressure [Pa]
P^S :	Pressure at the active layer/support surface [kPa]
q:	Molar loading [mol kg^{-1}]
q_M :	Molar saturation loading [mol kg^{-1}]
q_T :	Total molar loading [mol kg^{-1}]
Q:	Permeance [$\text{mol m}^{-2} \text{ s}^{-1} \text{ Pa}^{-1}$] or permeability [$\text{kg m}^{-2} \text{ h}^{-1}$]
$Q_{G,1}$:	Parameter defined by Eq. VI.40 [$\text{mol m}^{-2} \text{ s}^{-1} \text{ Pa}^{-1}$]
$Q_{G,2}$:	Parameter defined by Eq. VI.41 [$\text{m}^{-1} \text{ Pa}^{-1}$]
r:	Reaction rate [$\text{mol s}^{-1} \text{ kg}^{-1}$]; Radius [m]
R:	Constant of gases [$8.314 \text{ Pa m}^3 \text{ mol}^{-1} \text{ K}^{-1}$]; Radius [m]; Resistance offered by the membrane according to Eqs. VI.22-VI.24.
Re_p :	Particle Reynolds number [-]
S:	Surface [m^2]
\bar{S} :	Specific entropy [$\text{J mol}^{-1} \text{ K}^{-1}$]
S_{ext} :	External specific surface of a particle [$\text{m}^2 \text{ g}^{-1}$]; External surface [m^2]
S_{in} :	Internal surface [m^2]
S_r^2 :	Residual mean squares according to Eq. F.13
S_e^2 :	Error mean squares according to Eq. F.14
$S_{\bar{Y}_i}^2$:	Estimated variance of variable \bar{Y}_i
SWG:	Seeding weight gain [mg cm^{-2}]
t:	Time [s]
T:	Temperature [K]
T_b :	Boling point [K]

T^* :	Characteristic temperature in Eq. E.4
T_M :	Mean temperature of an experimental series [K]
u_o :	Surface velocity [m s^{-1}]
U :	Internal energy [J]
\bar{U} :	Specific internal energy [J kg^{-1}]
V :	Volume [m^3]
\bar{V} :	Molar volume [$\text{m}^3 \text{mol}^{-1}$] or specific volume [$\text{m}^3 \text{kg}^{-1}$]
\bar{V}_b :	Molar volume at boiling point [$\text{m}^3 \text{mol}^{-1}$]
w :	Molar flow [mol s^{-1}]
W :	Weight of zeolite NaA powder [kg]
X :	Weight fraction in the liquid phase [-]; Reaction conversion [-]
x :	Molar fraction in the liquid phase or in the adsorbate in Eq. I.30 [-]
Y :	Weight fraction in the gas phase [-]; Integral free energy relative to saturation loading [J kg^{-1}]
y :	Molar fraction in the gas phase [-]; Variable defined by Eqs. I.5 and VII.28
z :	Number of nearest neighbor sites [-]
Z :	Variable defined in Eq. VII.15
z_L :	Length of capillary filled with liquid [m]

Greek symbols:

α :	Characteristic parameter in a log-normal distribution [-] (Eq. VI.25); Exponent of the Dubinin-Astakhov isotherm (Eqs. I.14 and VII.16) [-]; Dimensionless flux (Eq. VIII.26) [-]
α_{ij} :	Selectivity towards the separation of species i from j [-]
β :	Parameter defined in the generalized Langmuir isotherm (Eqs. I.5 and VII.28) [-]; Molecular volume in Eq. I.11 [m^3]; Characteristic parameter in a log-normal distribution [-] (Eq. VI.25)
$\chi(E)$:	Distribution function of adsorption energies in Eqs. I.4 and VII.27 [-]
δ :	Parameter defined in the generalized Langmuir isotherm (Eqs. I.5 and I.6 and VII.28) [-].
δ_i :	Parameter defined in Eq. E.5
Δ :	Increment [-]
$\Delta\bar{G}^a$:	Integral free energy of the adsorbed phase [J kg^{-1}]
$\Delta\bar{g}^a$:	Differential free energy of the adsorbed phase [J mol^{-1}]
ΔH^o :	Adsorption enthalpy [J mol^{-1}]
ΔS^o :	Adsorption entropy [$\text{J mol}^{-1} \text{K}^{-1}$]
ε :	Porosity [-]
$\varepsilon(d)$:	Pore size density function
ε_i :	Parameter defined in Eq. E.6
ϕ_v :	Volumetric gas flow [$\text{cm}^3 \text{(STP) min}^{-1}$]

Φ :	Surface potential in the zeolite pores [J kg^{-1}]; Parameter defined in Eq. VI.11 [$\text{mol m}^{-4} \text{s}^{-1} \text{Pa}^{-1}$]
γ :	Activity coefficient [-]; Shear rate [s^{-1}]
γ_i^∞ :	Activity coefficient at infinite dilution
Γ :	Gamma function [-]
Γ_{ij} :	Thermodynamic factors between species i and j according to Eq. I.43
η :	Dimensionless z-coordinate [-]
φ :	Origin ordinate in Eq. VI.71 [$\text{mol m}^{-2} \text{s}^{-1} \text{Pa}^{-1}$],
K :	Adsorption constant [Pa^{-1}]
K_f :	Constant in Freundlich isotherm in Eqs. I.8
K_∞ :	Pre-exponential factor in Eqs. I.6
λ :	Mean-free path of the molecules [m]; Displacement of an adsorbed molecule in Eq. I.36 [m]; Slope in Eq. VI.70 [$\text{mol m}^{-2} \text{s}^{-1} \text{Pa}^{-2}$]
μ :	Chemical potential of the adsorbate [J mol^{-1}]; Viscosity [$\text{kg m}^{-1} \text{s}^{-1}$]
μ' :	Chemical potential of the adsorbent [J kg^{-1}]
μ_D :	Dipole moment [debye]
ν :	Jump frequency [s^{-1}]; Exponent in Eq. VI.14; Relative frequency [-]
π :	Spreading pressure [N m^{-1}]
Π :	Reduced pressure [-]
θ :	Fractional occupancy [-]; Contact angle [$^\circ$]; Time lag [s]
$\bar{\theta}$:	Mean fractional occupancy in a particle [-]
$\bar{\bar{\theta}}$:	Mean fractional occupancy in a bed of particles [-]
ρ :	Density [kg m^{-3}]
σ :	Surface tension [N m^{-1}]
σ_i :	Parameter defined in Eq. E.7 [\AA]
τ :	Tortuosity [-]; Shear stress [Pa]
Ω_D :	Collision integral [-]
ξ :	Variable defined in Eq. E.12
Ψ :	Parameter defined in Eq. VI.10 [$\text{mol m}^{-3} \text{s}^{-1} \text{Pa}^{-1}$]; Integral free energy relative to saturation loading (Eq. VII.15) [J mol^{-1}]
∇ :	Nabla operator [m^{-1}]

Subscripts:

ads:	Adsorption
b:	Bed
bl:	Blank
bulk:	Bulk
C:	Capillary

DNPE:	Di-n-pentylether
E:	Ethanol
eff:	Effective
eq:	Equilibrium
G:	Gas
in:	Inner
inter:	Intercrystalline
Kn:	Knudsen
L:	Liquid
M:	Membrane layer
m:	Mean; Molecule; Mixture
min:	Minimum
max:	Maximum
o:	Feed in a membrane module; Inlet
p:	Particle
1-pent:	1-pentanol
S:	Support or solid
surf:	Surface
T:	Total
v:	Vapor
w:	Water
V:	Vacancy
ZA:	Zeolite A

Superscripts:

a:	Adsorbate
C:	Capillary
D:	Diffusion
E:	Ethanol
in:	Inlet of the packed bed
Kn:	Knudsen
L:	Langmuir
m:	Molecule / solute
max:	Maximum
min:	Minimum
o:	Standard state or unary isotherm
out:	Outlet of the packed bed
P:	Permeate
R:	Retentate

S:	Surface or support
T:	Total
V:	Viscous
W:	Water

Acronyms

AFM:	Atomic force microscopy
BET:	Brunauer-Emmett-Teller isotherm
C:	Continuous
CF:	Centrifugal field
D:	Dialysis
DNPE:	Di-n-pentyl ether
DSC:	Differential scanning calorimetry
ED:	Electrodialysis
EDS:	Energy dispersive spectroscopy
EMD:	Equilibrium molecular dynamics
EMT:	External mass transfer
ESR:	Electron spin resonance
FESEM:	Field energy electron microscopy
GF:	Generalized Freundlich isotherm
GS:	Gas separation
GMS:	Generalized Maxwell-Stefan
HPLC:	High performance liquid chromatography
IAST:	Ideal Adsorbed Solution Theory
IMT:	Internal mass transfer
INN:	Inner-side membrane
LF:	Langmuir-Freundlich isotherm
LG:	Generalized Langmuir isotherm
MF:	Microfiltration
MS:	Maxwell-Stefan
NF :	Nanofiltration
NEMD:	Non-equilibrium molecular dynamics
NSE:	Neutron spin-echo
NMR:	Nuclear magnetic resonance
OBMC:	Orientalional-Bias Monte Carlo
OUT:	Outer-side membrane
PCS:	Photon correlation spectroscopy
PFG-NMR:	Pulse field gradient nuclear magnetic resonance
PRAST:	Predictive Real Adsorbed Solution Theory

PSD :	Pore size distribution
PTI:	Potential Thermodynamic Isotherm
PV:	Pervaporation
QENS:	Quasi-elastic neutron scattering
RAST:	Real Adsorbed Solution Theory
RO:	Reverse osmosis
SAC:	Steam-Assisted crystallization
SANS:	Small-angle neutron scattering
SBU:	Secondary building unit
SC:	Semi-continuous
SDA:	Structure directing agent
SEM:	Scanning electron microscopy
SGPV:	Sweep gas pervaporation
SQ:	Sum of squares
SSU:	Structural subunit
T:	Tóth isotherm
TEM:	Transmission electron microscopy
TEOM:	Tapered oscillating microbalance measurements
TGA:	Thermogravimetric analyzer
UF:	Ultrafiltration
UFDR:	Ultrasonic frequency domain reflectometry
VLE:	Vapor-liquid equilibrium
VP:	Vapor permeation
VPT:	Vapor Phase Transport
VPV:	Vacuum pervaporation
XRD:	X-ray diffraction
XRF:	X-ray fluorescence